

Balancing Chemical Equations

Name: _____

Directions: Balance the chemical statement. You may draw a representation of the balanced chemical reaction, do an accounting inventory or any other method you would like to use.

1) ___ $O_3 \rightarrow$ ___ O_2	2) ___ $KClO_3 \rightarrow$ ___ $KCl +$ ___ O_2
3) ___ $Al +$ ___ $O_2 \rightarrow$ ___ Al_2O_3	4) ___ $Cu +$ ___ $AgNO_3 \rightarrow$ ___ $Cu(NO_3)_2 +$ ___ Ag
5) ___ $C_3H_8 +$ ___ $O_2 \rightarrow$ ___ $CO_2 +$ ___ H_2O	

Balance the following:



