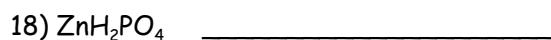
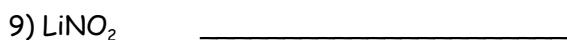
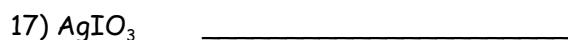
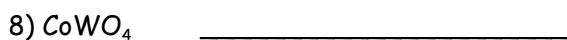
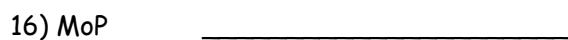
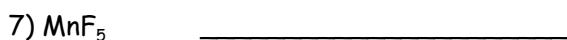
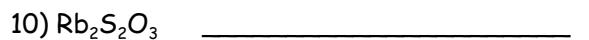


## Ionic Compounds #3

Name: \_\_\_\_\_

Part 1: Name the following ionic compounds.



Part 2: Write the formulas for the following compounds.

1) beryllium acetate \_\_\_\_\_

10) calcium fluoride \_\_\_\_\_

2) strontium arsenate \_\_\_\_\_

11) sodium oxalate \_\_\_\_\_

3) aluminum oxide \_\_\_\_\_

12) magnesium chloride \_\_\_\_\_

4) potassium bicarbonate \_\_\_\_\_

13) potassium hydroxide \_\_\_\_\_

5) ammonium sulfide \_\_\_\_\_

14) lead (IV) thiosulfate \_\_\_\_\_

6) iron (II) nitrite \_\_\_\_\_

15) magnesium silicate \_\_\_\_\_

7) nickel (III) fluoride \_\_\_\_\_

16) chromium (II) chloride \_\_\_\_\_

8) calcium bromide \_\_\_\_\_

17) indium (III) sulfide \_\_\_\_\_

9) ammonium phosphate \_\_\_\_\_

18) barium phosphate \_\_\_\_\_