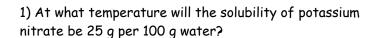
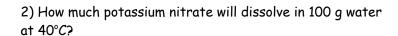
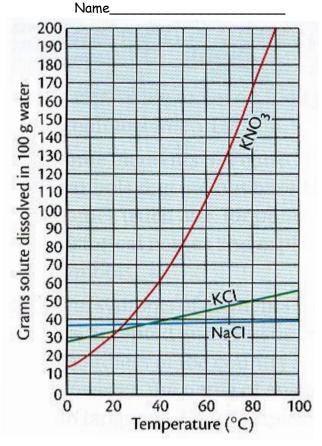
## Solubility Curves





- 3) What mass (in grams) of potassium nitrate (KNO $_3$ ) will dissolve in 100 g water at 60 °C?
- 4) What mass (in grams) of potassium chloride (KCl) will dissolve in 100 g water at 60  $^{\circ}$ C?
- 5) You dissolve 25 g potassium nitrate in 100 g water at 30  $^{\circ}C$ , producing an unsaturated solution. How much more potassium nitrate (in grams) must be added to form a saturated solution at 30  $^{\circ}C$ ?



- 6) What is the minimum mass (in grams) of 30  $^{\circ}$ C water needed to dissolve 25 g potassium nitrate?
- 7) You place 50 g NaCl in 100 g water at 30 °C.
- a) Classify the solution as unsaturated or saturated.
- b) Of the 50 g NaCl, about what mass will dissolve?
- c) Describe what you would see in the beaker.
- 8) At 80  $^{\circ}$ C, more KCl can dissolve in 100 g of water than at 10  $^{\circ}$ C. How many more grams of KCl can dissolve in the warmer water than the colder water?

9a) At what temperature could you dissolve 30 grams of potassium chloride in 100 grams of water?		
b) What temperature would you need 10 potassium nitrate?	00 grams of water to be at in or	der to dissolve 95 grams of
10a) At 90 $^{\circ}$ C, how many grams of sodiu	m chloride would you be able to	dissolve in 100 grams of water?
b) Suppose you doubled the amount of vable to be dissolved?	vater to 200 g. How many gram	s of sodium chloride would now be
c) Suppose you cut the amount of water able to dissolve now?	down to 50 grams. How many g	grams of sodium chloride would be
11) How many grams of KNO3 would dis	solve in 400 grams of 75 °C wat	er?
12) How many grams of KCl would dissol	ve in 25 grams of water at 10 $^\circ$ (	C?
	saturated, unsaturated or super 19 grams of NaCl in 100 grams water at 20 °C.	
14) What type of solution would you hav	ve if 90 grams of NaCl were diss	solved in 300 grams of water at