

Ionic Compounds #2

Name: _____

	Elements & Groups	Positive Ion	Negative Ion	How Many Positive Ions?	How Many Negative Ions?	Formula	Name
Ex	Barium & Nitrate	Ba^{+2}	NO_3^{-1}	1	2	$Ba(NO_3)_2$	Barium nitrate
1	Ammonium & Chlorine						
2	Potassium & Sulfate						
3	Cesium & Phosphate						
4	Sodium & Chlorate						
5	Magnesium & Acetate						
6	Lithium & Phosphorous						
7		NH_4^{+1}	NO_3^{-1}				
8		Ca^{+2}	SO_4^{-2}				
9		Cr^{+3}	$C_2O_4^{-2}$				

	Elements	Positive Ion	Negative Ion	How Many Positive Ions?	How Many Negative Ions?	Formula	Name
10		Sn^{+4}	Cl^{-1}				
11							Potassium perchlorate
12							Aluminum hydroxide
13							Hydrogen chromate
14							Potassium permanganate
15							Copper (II) sulfate
16						$\text{Zn}(\text{BrO}_3)_2$	
17						Na_2S	
18						AgNO_3	
19						$(\text{NH}_4)_2\text{O}$	
20						$\text{Sr}(\text{C}_2\text{H}_3\text{O}_2)_2$	