

## Unit 0 Quiz Review

Name: \_\_\_\_\_

1a) Define quantitative data:

Data that gives a number description of an observation

b) Describe your kitchen in a quantitative way:

There are 20 cabinets in my kitchen

2a) Define qualitative observation:

Data that gives a description of an observation that does not involve a number

b) Describe your kitchen in a qualitative way:

My kitchen has brown cabinets and a white countertop.

3) Measuring

Mass	Volume	Length	Substance
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- A. A meter measures? length
- B. A gram measures? mass
- C. What does a mole measure? substance
- D. What does a liter measure? volume

4) How do you convert?

- A. Cm to m: move decimal 2 places to the left
- B. Mm to cm move decimal 8 places to the right
- C. Decimeters to meters: move decimal 1 place to the left
- D. liters to microliters: move decimal 6 places to the right
- E. Grams to mega grams: move decimal 6 places to the left
- F. Kg to grams: move decimal 3 places to the right
- G. Meters to decimeters: move decimal 1 place to the right

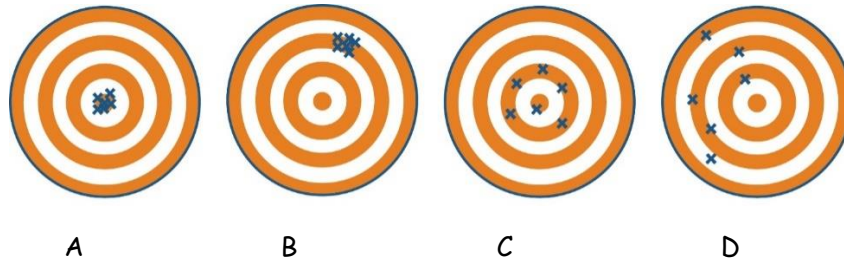
5) Convert the following:

a) 160 cg = 0.00160 kg

c) 0.00054 hs = 0.54 ds

b) 0.0078 m = 7800  $\mu\text{m}$

d) 19000 mL = 19 L



6) Which picture is:

- a. Accurate and not precise C
- b. Precise but not accurate B
- c. Neither accurate nor precise D
- d. Both accurate and precise A

7) What is density?

Density is a measure of how close the particles of a substance are to one another. It is calculated by taking the mass divided by the volume of the object. A low density means the particles are spread out.

8a) What is the density of a substance that has a mass of 33.4 grams and a volume of 5 cm<sup>3</sup>? Remember to include units!

$$6.7 \text{ g/cm}^3$$

b) A substance has a density of 6.1 g/mL and a volume of 3.4 mL. What is the mass of the substance? Include units!

$$20.8 \text{ g}$$

c) A student performs an experiment with three unknown fluids and obtains the following measurements:

Fluid A:  $m = 315 \text{ g}$   $V = 200 \text{ mL}$   $D = 1.6 \text{ g/mL}$

Fluid B:  $m = 268 \text{ g}$ ,  $V = 112.5 \text{ mL}$   $D = 2.4 \text{ g/mL}$

Fluid C:  $m = 147.5 \text{ g}$ ,  $V = 375 \text{ mL}$   $D = 0.47 \text{ g/mL}$

Least Dense: Fluid C

Dense: Fluid A

Most Dense: Fluid B

d) Fluids A-C in the above question are mixed and allowed to settle out. Which one will float on top, which will be in the middle, and which will sink to the bottom?

Fluid C would float on top, then Fluid A, then Fluid B would sink to the bottom